# 9(R)-[6(R)-Hydroxymethyl-1-oxa-4-thiacyclohexan-2-yl]hypoxanthinewater (4/3), a nucleoside analogue 

G. Mazumder, ${ }^{\text {a }}$ M. De, ${ }^{\text {a* }}$ S. K. Mazumder, ${ }^{\text {a }}$<br>A. Mukhopadhyay, ${ }^{\text {b }}$ A. K. Das ${ }^{\text {c }}$ and A. Van Aerschot ${ }^{\text {d }}$

${ }^{\text {a D Department of Materials Science, Indian Association for the Cultivation of Science, }}$ Jadavpur, Calcutta 700 032, India, ${ }^{\mathbf{b}}$ Department of Physics, University of Jadavpur, Jadavpur, Calcutta 700 032, India, ${ }^{\text {'Biotechnology Centre, Indian Institute of }}$ Technology, Kharagpur 721 302, India, and daboratorium Voor Farmaceutesche Chemie, Rega Institute of Medical Research, Katholieke Universiteit, Leuvan, B-3000 Leuvan, Belgium
Correspondence e-mail: msmd@mahendra.iacs.res.in
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The title compound, $\quad 9(R)$-[6(R)-hydroxymethyl-1-oxa-4-thiacyclohexan-2-yl]-1,9-dihydro-6 H -purin-6-one-water (4/3), $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{~S} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}$, crystallizes in the triclinic space group $P 1$ with four molecules in the asymmetric unit and 0.75 waters of hydration per molecule. The structure was refined to an $R$ value of 0.072 for 3382 observed reflections. The four crystallographically independent molecules are designated $A, B, C$ and $D$. All four oxathiane rings adopt the chair conformation and the purine bases are in an anti orientation with respect to the sugar moieties. Molecules $A$ and $D$ and molecules $C$ and $B$ are base paired by a single hydrogen bond of the type $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$. These base pairs are again hydrogen bonded to their translated pairs in the direction of a cell diagonal.

## Comment

A number of sugar-modified nucleosides display antiviral activity because of their ability to interfere with virally encoded enzymes. The structure analysis of the title compound, (I), is a continuation of our earlier work involving systematic conformational studies on modified nucleosides with the aim of understanding structure-function relationships. The synthesis of (I) has been described (Van Aerschot et al., 1990).

The crystal structure of (I) (Fig. 1) has been determined and the absolute configuration is indicated by a Flack parameter of 0.04 (9) (Flack, 1983). There are four crystallographically independent molecules in the crystal structure, designated $A$, $B, C$ and $D$. All bond lengths and angles of the four hypoxanthine residues agree well with the values quoted by Taylor and others (Taylor \& Kennard, 1982). The C6-O6 bond
lengths in the four molecules vary from 1.204 (6) to 1.242 (6) $\AA$, with an average value of 1.217 (6) $\AA$, which agrees well with the quoted value of $1.215 \AA$ for $\mathrm{C}=\mathrm{O}$ (Taylor \& Kennard, 1982). The pyrimidine and imidazole rings in the four molecules are planar within experimental error. The dihedral angles between pyrimidine and imidazole rings are 0.7 (2), 2.1 (1), 1.3 (2) and $0.2(2)^{\circ}$ in molecules $A, B, C$ and $D$, respectively. The bond lengths and angles of the sugar moieties of the four molecules agree well with each other within experimental error. The N -glycosidic bond lengths

(I)
$\mathrm{N} 9-\mathrm{C}^{\prime}$ are in the range $1.443(6)-1.465(6) \mathrm{A}$, with an average value of 1.451 (6) $\AA$, which agrees well with the normal value of 1.464 (6) $\AA$. In all four molecules, the $\mathrm{C}^{\prime}-\mathrm{O}^{\prime}$ bond lengths lie in the range $1.402(5)-1.422(5) \AA$, with an average value of 1.413 (5) $\AA$. This is shorter than the adjacent bond lengths $\mathrm{C}^{\prime}-\mathrm{O}^{\prime}$, which lie in the range $1.420(5)-$ 1.444 (5) $\AA$, with an average value of 1.430 (6) $\AA$. Similar shortening of bond lengths has been found in glucosides with an equatorial substituent at the anomeric C atom $\mathrm{C}^{\prime}$ (Berman et al., 1967; Jeffrey et al., 1978). This anomeric effect is commonly observed in other nucleosides (Birnbaum et al., 1984; Thewalt et al., 1970; Watson et al., 1965). The orientation of $\mathrm{O}^{\prime}$ with respect to the sugar moiety described by the torsion angle $\gamma\left(\mathrm{O} 7^{\prime}-\mathrm{C} 7^{\prime}-\mathrm{C} 6^{\prime}-\mathrm{C} 5^{\prime}\right)$ is different for all four molecules. In molecules $A, B$ and $D$, the $\gamma$ values are -168.3 (5), 175.4 (5) and -179.3 (6) ${ }^{\circ}$, respectively, and they are in the gauche-trans ( ap ) orientation, whereas molecule $C$ has $\gamma=-70.2(6)^{\circ}$, which corresponds to a trans-gauche $(-s c)$


Figure 1
An ORTEPII (Johnson, 1976) plot of one of the molecules shown with 30\% probability ellipsoids.
orientation. The puckering parameters (Cremer \& Pople, 1975) for the sequence $\mathrm{C} 2^{\prime}-\mathrm{C} 3^{\prime}-\mathrm{S} 4^{\prime}-\mathrm{C}^{\prime}-\mathrm{C} 6^{\prime}-\mathrm{O} 1^{\prime}$ in the oxathiane ring are $Q=0.631(5) \AA, \varphi_{2}=-134(5)^{\circ}$ and $\theta_{2}=$ $175.2(5)^{\circ}$ for molecule $A, Q=0.629(5) \AA, \varphi_{2}=-71(4)^{\circ}$ and $\theta_{2}=172.9(4)^{\circ}$ for molecule $B, Q=0.623(6) \AA, \varphi_{2}=-44(6)^{\circ}$ and $\theta_{2}=173.9(5)^{\circ}$ for molecule $C$, and $Q=0.663$ (8) $\AA, \varphi_{2}=$ $-117(3)^{\circ}$ and $\theta_{2}=170.1(7)^{\circ}$ for molecule $D$, giving an average value of $Q=0.639$ (6) $\AA$ for the total puckering amplitude; this value is close to that of $0.63 \AA$ for an ideal cyclohexane ring (Cremer \& Pople, 1975). Each molecule of the title compound has two H atoms capable of participating in hydrogen bonds and these are from N 1 of the purine bases and the hydroxyl $\mathrm{O}^{\prime}$ atom of the sugar ring. There are hydrogen bonds between purine rings. The purine rings form $A D$ and $B C$ base pairs by a single hydrogen bond of the type $\mathrm{N} 1-\mathrm{H} \cdots \mathrm{N} 7$ and these base pairs are hydrogen bonded to their translated pairs in the direction of a cell diagonal. The bases in each pair are nearly parallel to each other, the dihedral angles between them being $170.9(2)^{\circ}$ for the $A D$ and 171.1 (1) ${ }^{\circ}$ for the $B C$ base pairs.

## Experimental

The synthesis of the title compound has been described by Van Aerschot et al. (1990). Colorless prismatic crystals were grown by slow evaporation from methanol at room temperature ( 293 K ).

## Crystal data

$\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{~S} \cdot 0.75 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=281.81$
Triclinic, $P 1$
$a=7.995(2) \AA$
$b=9.261(2) \AA$
$c=18.062(2) \AA$
$\alpha=98.290(10)^{\circ}$
$\beta=94.470(10)^{\circ}$
$\gamma=101.210(10)^{\circ}$
$V=1290.4(4) \AA^{\circ}$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.451 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 25 \\
& \quad \text { reflections } \\
& \theta=9.6-14.4^{\circ} \\
& \mu=0.265 \mathrm{~mm}^{-1} \\
& T=293(2) \mathrm{K} \\
& \text { Prism, colorless } \\
& 0.25 \times 0.21 \times 0.12 \mathrm{~mm}
\end{aligned}
$$

Data collection
Enraf-Nonius CAD-4 diffractometer
$\omega-2 \theta$ scans
Absorption correction: empirical via $\psi$ scan (North et al., 1968)
$T_{\text {min }}=0.848, T_{\text {max }}=0.973$
4559 measured reflections
4559 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.072$
$w R\left(F^{2}\right)=0.155$
$S=0.958$
4559 reflections
673 parameters
H atoms treated by a mixture of independent and constrained refinement

Table 1
Hydrogen-bonding geometry ( $\AA,^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | D-H | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 1 W-\mathrm{H} 1 W 1 \cdots \mathrm{O} 2 W$ | 0.83 | 2.07 | 2.804 (7) | 148 |
| $\mathrm{O} 1 W-\mathrm{H} 2 W 1 \cdots \mathrm{~N} 3 B$ | 0.84 | 2.06 | 2.901 (6) | 174 |
| $\mathrm{N} 1 A-\mathrm{H} 1 A \cdots \mathrm{~N} 7 D^{\mathrm{i}}$ | 0.86 (6) | 1.96 (6) | 2.816 (7) | 176 (4) |
| $\mathrm{O}^{\prime} A-\mathrm{H} 7^{\prime} A \cdots \mathrm{O} 1 W^{\mathrm{ji}}$ | 0.99 | 1.78 | 2.734 (7) | 163 |
| $\mathrm{O}^{\prime} C-\mathrm{H}^{\prime} C \cdots \mathrm{O}^{\prime} A^{\text {iii }}$ | 0.83 | 2.20 | 2.769 (8) | 126 |
| $\mathrm{N} 1 B-\mathrm{H} 1 B \cdots \mathrm{~N} 7 C^{\text {iv }}$ | 0.86 (5) | 2.00 (6) | 2.858 (6) | 172 (4) |
| $\mathrm{O} 2 W-\mathrm{H} 2 W 2 \cdots \mathrm{O} 6 \mathrm{C}^{\mathrm{v}}$ | 1.05 | 1.95 | 2.753 (7) | 130 |
| $\mathrm{O}^{\prime} B-\mathrm{H} 7^{\prime} B \cdots \mathrm{O} 6 D^{\text {vi }}$ | 0.84 | 1.92 | 2.728 (6) | 162 |
| $\mathrm{N} 1 C-\mathrm{H} 1 C \cdots \mathrm{~N} 7 B^{\text {vii }}$ | 0.86 (6) | 2.02 (6) | 2.882 (6) | 178 (4) |
| $\mathrm{N} 1 D-\mathrm{H} 1 D \cdots \mathrm{~N} 7 A^{\text {viii }}$ | 0.86 (6) | 1.97 (6) | 2.829 (7) | 174 (5) |
| $\mathrm{O}^{\prime} D-\mathrm{H} 7^{\prime} D \cdots \mathrm{O} 6 B^{\text {viii }}$ | 0.83 | 2.14 | 2.802 (8) | 137 |
| $\mathrm{O} 2 W-\mathrm{H} 1 W 2 \cdots \mathrm{O}^{\prime} B^{\text {ix }}$ | 0.82 | 2.20 | 2.938 (8) | 150 |
| $\mathrm{O} 3 W-\mathrm{H} 1 W 3 \cdots \mathrm{O} 6 A^{\mathrm{x}}$ | 0.85 | 2.00 | 2.833 (7) | 167 |
| $\mathrm{O} 3 W-\mathrm{H} 2 W 3 \cdots \mathrm{O} 7^{\prime} D^{\text {xi }}$ | 0.91 | 1.89 | 2.793 (9) | 167 |

Symmetry codes: (i) $x-1, y-1, z-1$; (ii) $x-1, y, z$; (iii) $1+x, 1+y, z$; (iv) $x, y-1, z-1$; (v) $x, y, z$; (vi) $x, y, z-1$; (vii) $x-1, y, 1+z$; (viii) $x, y, 1+z$; (ix) $x, y-1, z ;$ (x) $x, 1+y, 1+z ;$ (xi) $x-1,1+y, z$.

H atoms of hydroxyl groups and water molecules were located from a difference map and not refined. Other H atoms were obtained from stereochemical considerations and were refined with fixed isotropic displacement parameters $(\mathrm{C}-\mathrm{H}=0.93-0.98 \AA$ ).

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: CAD-4 Software; program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s) used to refine structure: SHELXL93 (Sheldrick, 1993); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL93.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1134). Services for accessing these data are described at the back of the journal.

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